

An Example of Lawvere's
Entropy Framework
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Applied Category Theory
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Lawvere's Framework

Lawvere postulates a category \mathcal{P} wherein:

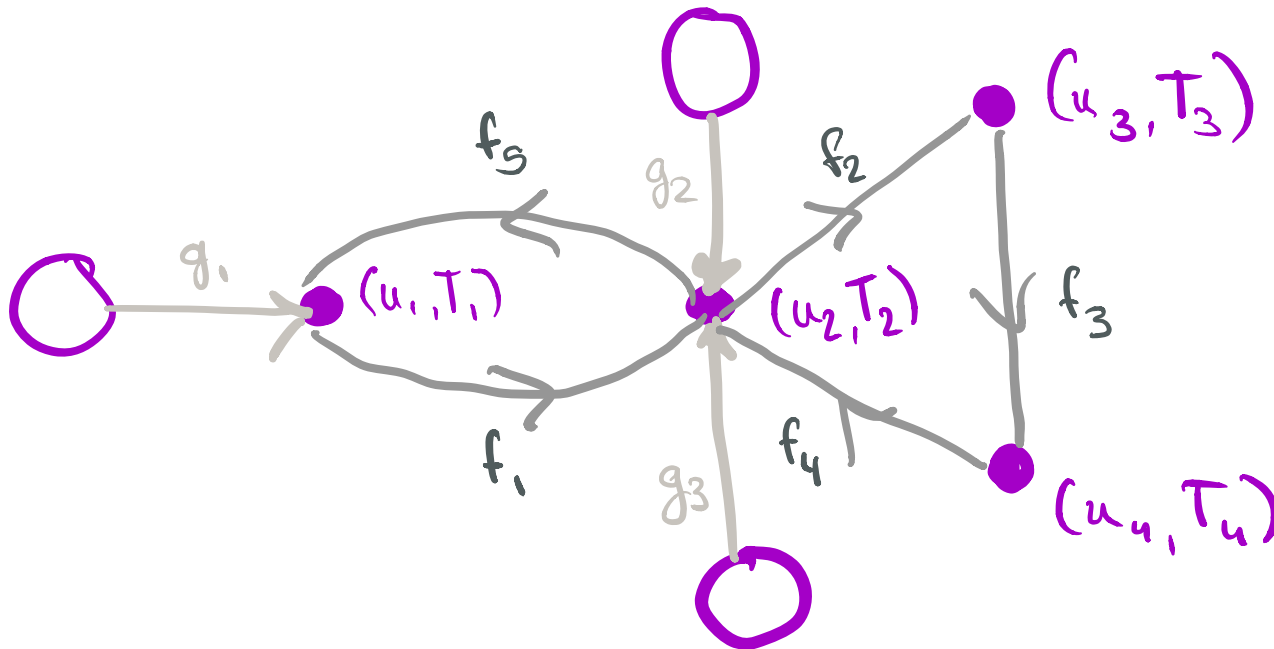
- $\text{ob}(\mathcal{P})$ is the collection of states of the thermodynamic system
- $\text{ar}(\mathcal{P})$ is the collection of physical processes

Lawvere assumes there is given a functor from \mathcal{P} to the additive monoid of extended reals, which he calls the entropy supply, called Δ .

processes either consume (+) or generate (-) entropy.

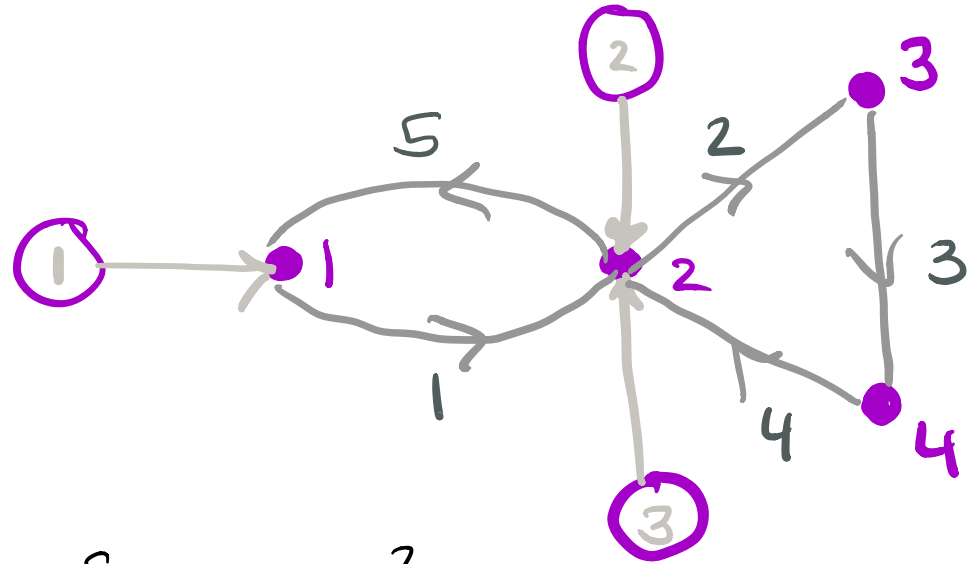
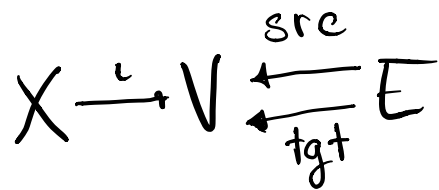
Heat Networks

Lawvere doesn't provide an example of \mathcal{P} or \mathcal{S} , so we set out to develop one.



Entropy supply will be related to flows through the interfaces of such a network.

Open Graphs as Heat Networks



$$X = \{1, 2, 3\}$$

$$1 \xrightarrow{c} 1 \quad 2 \xrightarrow{c} 2$$

$$3 \xrightarrow{c} 2$$

$$V = \{1, 2, 3, 4\}$$

$$E = \{1, 2, 3, 4, 5\}$$

$$1 \xrightarrow{(\text{src}, \text{tgt})} (1, 2)$$

$$3 \xrightarrow{(\text{src}, \text{tgt})} (3, 4)$$

$$5 \xrightarrow{(\text{src}, \text{tgt})} (2, 1)$$

$$2 \xrightarrow{(\text{src}, \text{tgt})} (2, 3)$$

$$4 \xrightarrow{(\text{src}, \text{tgt})} (4, 2)$$

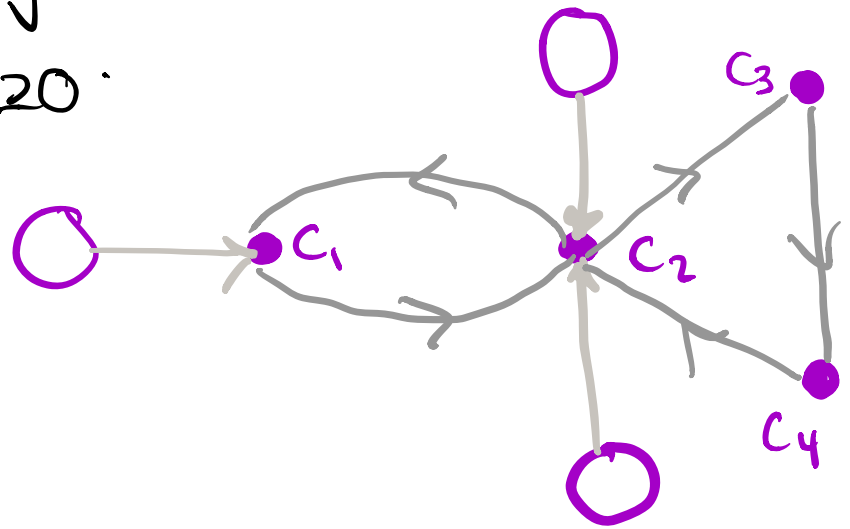
Open Graphs as Heat Networks, II

The open graph is equipped with a function $C: V \rightarrow \mathbb{R}_{\geq 0}$ that assigns a **heat capacity** to each vertex.

A **state** of the heat network is specified by a function $u: V \rightarrow \mathbb{R}_{\geq 0}$ that assigns an **energy** to each vertex.

The **phase space** is $\mathbb{R}_{\geq 0}^V$.

The **temperature** of a vertex $v \in V$ at a certain state is $T_v = u_v / c_v$.



Processes in Heat Networks

A **process** consists of:

- a time interval $[0, \tau]$

- a piecewise-differentiable function

$U: [0, \tau] \rightarrow \mathbb{R}_{\geq 0}^V$ assigning **internal energy**

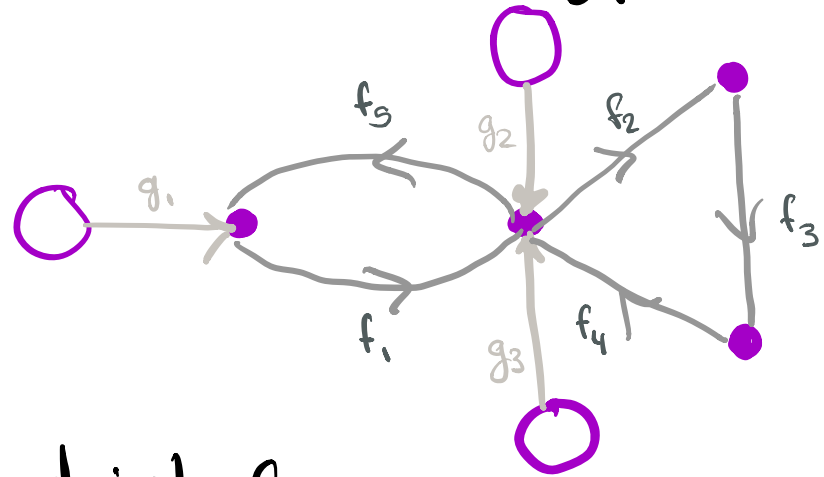
- integrable functions

$$f: [0, \tau] \rightarrow \mathbb{R}^E$$

$$g: [0, \tau] \rightarrow \mathbb{R}^X$$

assigning **heat fluxes**

to the internal edges and interfaces



such that...

Constraints on Processes

$\forall v \in V$. almost all $t \in [0, \tau]$.

$$\bullet U'_v(t) = \sum_{t \text{gt}(e)=v} f_e(t) - \sum_{\text{src}(e)=v} f_e(t) + \sum_{i(x)=v} g_x(t)$$

Energy is conserved.

$$\bullet f_e(t) \geq 0 \text{ if } T_{\text{src}(e)} > T_{\text{trg}(e)}$$

$$f_e(t) \leq 0 \text{ if } T_{\text{src}(e)} < T_{\text{trg}(e)}$$

Heat only flows from hot to cold.

Category of Heat Network Processes

Define a category **HeatProc**:

- $\text{ob}(\text{HeatProc})$ is the collection of states

$$u \in \mathbb{R}_{\geq 0}^V$$

- $\text{ar}(\text{HeatProc})$ is the collection of processes.

The domain of a process is the initial state and the codomain is the final state.

- identity morphisms are "freeze" processes, which have duration zero and zero heat flow
- composition of processes is concatenation



Entropy Supply Functor

Lawvere assumes an entropy supply functor

$$\mathcal{S} : \mathcal{P} \rightarrow \mathcal{B}(\mathbb{R}, +).$$

In thermodynamics, $(\delta S = \frac{\delta Q}{T}) \Rightarrow (\Delta S = \int \frac{\delta Q}{T})$

We define the entropy supply functor for a process P

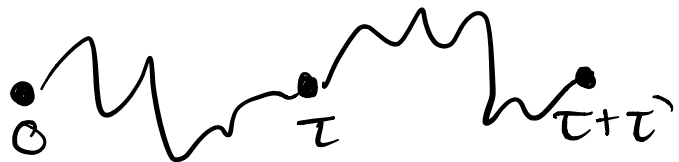
in HeatProc as $\mathcal{S}(P) := \int_0^\tau \sum_{x \in X} \frac{g_x(t)}{T_{L(x)}(t)} dt$

- \mathcal{S} preserves identity morphisms

$$\mathcal{S}(\text{id}_u) = 0$$

- \mathcal{S} preserves composition

$$\mathcal{S}(P \cdot P') = \mathcal{S}(P) + \mathcal{S}(P')$$



Naive Clausius Property

The Clausius theorem is $\oint \frac{\delta Q}{T_{\text{surr}}} \leq 0$, which is the statement that thermodynamic cycles cannot decrease entropy.

Lawvere defines the naive Clausius property for entropy supplies to be $\downarrow (P: u \rightarrow u) \leq 0$.

Naive Clausius Property, II

Theorem. $\Delta: \text{HeatProc} \rightarrow \mathcal{B}(\mathbb{R}, +)$ satisfies the naive Clausius property.

$$\begin{aligned} \Delta(P) &= \int_0^\tau \sum_{x \in X} \frac{g_x(t)}{T_{(x)}(t)} dt \\ &= \int_0^\tau \underbrace{\sum_{v \in V} \frac{U_v(t)}{T_v(t)} dt}_{=0 \text{ (cyclicality)}} - \int_0^\tau \sum_e f_e(t) \underbrace{\left(\frac{1}{T_{\text{tgt}(e)}(t)} - \frac{1}{T_{\text{src}(e)}(t)} \right)}_{\leq 0 \text{ (heat flows from hot to cold)}} dt \end{aligned}$$

(conservation of energy)

$$\Delta(P: u \rightarrow u) \leq 0 \quad \checkmark$$

Future Directions

- adapting network model to other thermodynamic fluxes (work, particles)
- investigating functoriality of construction of HeatProc from open graphs